Regulation of Intracellular pH in Human Melanoma: Potential Therapeutic Implications

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Abstract
Melanoma cells in vivo maintain intracellular pH (pHi) in a viable range despite an extracellular tumor pH (pHe) that is typically below 7.0. In general, three families of transporters are capable of removing metabolic protons, but the specific transporters responsible for the maintenance of pHi at low pHe in melanomas have not been identified. Although the transporters exist in most cells, an inhibitor would be predicted to have selectivity for cells located in an acidic tumor bed because cells in that environment would be expected to have transporters chronically activated. In this report, the levels and extent of expression of the Na\(^+\)/H\(^+\) exchanger (NHE-1) and two of the H\(^+\)-linked monocarboxylate transporters (MCTs) were evaluated in three melanoma cell lines. The effects of inhibitors of each transporter were tested at an extracellular pH (pHe) of 7.3, 6.7, or 6.5 in melanoma cells that were grown at pHe 7.3 or 6.7. The activity of MCT isoform 1 (MCT-1) was up-regulated in three melanoma cell lines at low pHe, but that of NHE-1 was not. Furthermore, NHE-1 activity was lower in the melanomas than in other normal and malignant cell lines that were tested. Reverse transcription-PCR using primers specific for MCT-1, MCT-4, and NHE-1 showed that expression of none of these transporters was reproducibly up-regulated at the level of transcription when cells were grown at pHe 6.7 instead of pHe 7.3. Ex vivo experiments using DB-1 human melanoma xenografts grown in severe combined immunodeficient mice found that MCT-1 and not NHE-1 was a major determinant of DB-1 tumor cell pHi. Taken together, the data indicate that MCTs are major determinants of pH regulation in melanoma. In contrast, keratinocytes and melanocytes under low pHe conditions relied on NHE-1. Inhibitors of MCTs thus have great potential to improve the effectiveness of chemotherapeutic drugs that work best at low pHe, such as alkylating agents and platinum-containing compounds, and they should be selective for cells in an acidic tumor bed. In most tissues, it is proposed that the NHE-1 could compensate for an inhibited MCT to prevent acidification, but in melanoma cells this did not occur. Therefore, MCT inhibitors may be particularly effective against malignant melanoma.

Introduction
Many cellular processes, such as enzyme activity, molecular synthesis, transport of metabolites, and cell cycle progression, are dependent on the maintenance of pHi in the physiological range. In tumors, despite an pHe of 5.6–7.6, the pH i is typically in the normal range of 6.9–7.4 (1). There has been interest in recent years in potentiating tumor therapy by compromising cellular pHi control using inhibitors of ion transport (2–5).

We have evaluated three classes of transporters used by the cell to regulate pHi. The NHE is found in most cell types that have been studied (6–10), including human melanoma cells (11). NHE-1 activity was evaluated in the present study with the inhibitor cariporide mesilate (HOE642) (12). The activity of the Na\(^{+}\)-dependent Cl\(^-/\)HCO\(_3\)\(^-\) exchanger was investigated with the inhibitor DIDS (13, 14). MCTs, which transport lactate or pyruvate together with a H\(^+\) in the same direction (symport), were also evaluated. MCT-1 is the most prevalent of these in tissues, especially in heart and red muscle, and is up-regulated by increased work (15). MCT-4 has been shown to be up-regulated in tissues that have a high glycolytic rate, such as white muscle fibers (16, 17). In the present study, the relative contribution of MCT-1 to pHi regulation was evaluated, using CNCn or DBDS (18, 19). Characteristics of these inhibitors and transporters are summarized in Table 1.

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3 The abbreviations used are: pH, intracellular pH; pHe, extracellular tumor pH; CNCn, α-cyano-4-hydroxycinnamate; BCECF, bis-carboxyethyl-carboxyfluorescein; C.SNARF, carboxy-ssemi-naphthyl-rhodofluor; DBDS, 4,4'-diisothiocyanato-stilbene-2,2'-disulfonic acid; DIDS, 4,4'-disothiocyanato-stilbene-2,2'-disulfonic acid; HMVEC, human microvascular endothelial cell; HUVEC, human umbilical vein endothelial cell; MCT, H\(^+\)-linked monocarboxylate transporter; NHE, Na\(^+/\)H\(^+\) exchanger (sodium-proton exchanger); NHEM, normal human epidermal melanocyte; RT-PCR, reverse transcription-PCR; SCID, severe combined immunodeficiency/immunodeficient; AM, acetoxymethyl ester; RGD, arginine-glycine-aspartic-aspatic acid; CCD, charge-coupled device; FACS, fluorescence-activated cell sorting; GAPDH, glyceraldehyde-3-phosphate dehydrogenase.

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In common with most cell lines that have been tested, the three melanoma lines showed very little sensitivity to any inhibitor of proton translocation when they were tested in a medium of pH\textsubscript{r} 7.3. Under conditions of extracellular acidification, however, all three showed a distinctive and striking up-regulation of one or more isoforms of MCT at the protein level. Melanoma cells exhibited little NHE-1 activity and no up-regulation of the MCT-1 transporter. The importance of this MCT to melanoma cell maintenance of pH\textsubscript{i} was confirmed by observations that melanoma cells that were grown under conditions of chronic acidification (pH\textsubscript{e} 6.7) were even more dependent on MCTs. In most cases, they exhibited an up-regulation of one or more isoforms of MCT at the protein level. Melanoma cells exhibited little NHE-1 activity and no up-regulation at pH\textsubscript{e} 6.7. In contrast, keratinocytes, frequently the cell type surrounding a melanoma in situ, had relatively higher NHE-1 and less MCT activity. Taken together, these results suggest that inhibitors of MCT activity may be useful adjuncts to melanoma therapy, and may be particularly active in tumor populations in which the pH\textsubscript{r} is lowest, typically areas that are most refractory to treatment.

### Materials and Methods

**Reagents.** The fluorochromes C.SNARF-1-AM and BCECF-AM and the detergent Pluronic F127 were obtained from Molecular Probes (Eugene, OR). Cariporide mesilate was the generous gift of Dr. Udo Albus (Hoechst AG, Cardiovascular Research, Frankfurt/Mainz, Germany). Type I rat tail collagen was obtained from Collaborative Biomedical Products (Bedford, MA). Fibronectin-like engineered protein polymer and all other reagents were obtained from Sigma Chemical Co. (St. Louis, MO) unless otherwise noted.

**Cell Culture.** The Sk-Mel-28 human melanoma cell line was obtained from the American Type Culture Collection (Rockville, MD). DB-1 and DB-8 melanoma cells were early passage human melanoma cells derived from lymph node biopsies from metastatic patients and were provided by Dr. David Berd (Thomas Jefferson University Hospital, Philadelphia, PA). All of these cell lines were maintained in logarithmic growth at 37°C as monolayers in 75-cm\textsuperscript{2} flasks. The melanoma cells were grown in α-modified Eagle’s medium (αMEM) containing 12 mM glucose, 2 mM nonessential amino acids, 2 mM glutamine, and 10% heat-inactivated fetal bovine serum (complete medium). This medium had a bicarbonate concentration of 26 mM, and it provided a pH\textsubscript{e} of 7.3 under 5% CO\textsubscript{2} and a pH\textsubscript{i} 6.7 under 17% CO\textsubscript{2}. Cells were subcultured every 7 days by trypsinization. The pH of the medium was stable over this time interval and was monitored regularly. Sk-Mel-28 cells had doubling times of 72 h, and the DB-1 and DB-8 cells had doubling times of 48 h. These doubling times were unchanged by growth at low pH\textsubscript{r} when achieved, as in the present study, by raising the CO\textsubscript{2} from 5% to 17% rather than reducing the bicarbonate in the medium. HaCaT cells, which are immortalized nontumorigenic keratinocytes, were provided by Norbert Fusenig (DKFZ, Heidelberg, Germany) and were grown in Madin-Darby canine kidney (MDCK) medium as described previously (20). Primary NHEMs were cultured by Dr. Ulrich Roedec (Department of Dermatology, Thomas Jefferson University) as described previously (21). Experiments with melanocytes and HaCaT cells were performed in medium devoid of growth factors. HMT3522-T4/2 malignant breast cells were provided by Pierriand (The Danish Cancer Society, Copenhagen, Denmark), and were maintained as described previously (5). PS120 cells, which overexpress NHE-1 protein (22), were provided by Diane Barber (University of California-San Francisco, San Francisco, CA) with kind permission from Jacques Pouyssegur (University of Nice, Nice, France).

HUVeCs were obtained from umbilical cord blood made available from the maternity ward of Thomas Jefferson Hospital and prepared as described previously (23). HUVeCs were cultured in M199 medium (Life Technologies, Inc., Rockville, MD), supplemented with 10% defined calf serum, 0.05% endothelial cell growth supplement (ECGS), 0.5% L-glutamine, 0.013% heparin sulfate, 0.5% penicillin/streptomycin, 0.05% gentamicin, and 0.5% fungizone. This medium also has a bicarbonate concentration of 26 mM and provided a pH\textsubscript{e} 7.3 under 5% CO\textsubscript{2} and a pH\textsubscript{i} 6.7 under 17% CO\textsubscript{2}. HUVECs dermal adult cells were obtained from Clonetics (Walkersville, MD) and were maintained in Clonetics’ medium.

**Preparation of Cells for pH\textsubscript{r} Experiments.** For determination of pH\textsubscript{r}, cells were plated on 35-mm microwell plastic dishes with glass coverslips glued to 1-cm diameter holes in the center of each dish (Mattek Corp., Ashland, MA). For melanoma cell experiments, coverslips were coated with a mixture of collagen I and RGD repeating peptides (Sigma) to provide a physiological substrate. RGD repeating peptides were used as a substitute for fibronectin. Collagen I and RGD peptides were both used at a final concentration of 50 μg/ml (24, 25). For experiments with HMT3522-T4/2 cells, coverslips were coated with a 1:3 dilution of Matrigel and serum-free medium as described previously (5). For experiments with NHEMs, coverslips were coated with fibronectin-like RGD fragments because fibronectin adhesion is thought to be critical to focal contact organization (26) and suppression of apoptosis in these cells (27). Cells were plated in the

### Table 1: Summary of inhibitors of ion transport used in the present study

<table>
<thead>
<tr>
<th>Inhibitor</th>
<th>Target</th>
<th>Notes</th>
<th>References</th>
</tr>
</thead>
<tbody>
<tr>
<td>CNCn</td>
<td>MCT-1, MCT-4</td>
<td>Reduces pH\textsubscript{i} in vitro</td>
<td>(16, 18, 38, 58, 59)</td>
</tr>
<tr>
<td>DBDS</td>
<td>MCT-1, MCT-4</td>
<td>Binds irreversibly, is a DIDS analogue, no data in vivo</td>
<td>(59)</td>
</tr>
<tr>
<td>DIDS</td>
<td>Cl\textsuperscript{-}/HCO\textsubscript{3}\textsuperscript{-} exchanger at 0.2 mM has been reported to inhibit MCT-1, MCT-4 at 2 mM</td>
<td>Binds irreversibly to the Cl\textsuperscript{-}/HCO\textsubscript{3}\textsuperscript{-} exchanger</td>
<td>(13, 14)</td>
</tr>
<tr>
<td>Cariporide mesilate (HOE642)</td>
<td>NHE-1</td>
<td>Water soluble</td>
<td>(12)</td>
</tr>
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[H11002](http://example.com)
microwell dishes at a density of 0.2–0.3 × 10⁶ cells/ml in 2 ml of medium, 48–72 h before experiments were performed. Preliminary experiments (not shown) found that dye leakage over 1–2 h of an experiment was minimal (usually undetectable) if the cells were previously plated 48–72 h before the experiment and allowed to adjust to their substrate.

**Dye Loading.** Cells were incubated for 15 min with 9 μM C.SNARF-1-a.m. (28) or for 4 min with 5 μM BCECF-a.m. in medium containing 10% fetal bovine serum in a 37°C incubator under 5% CO₂ as described previously (5, 29). After a change of medium, the cells were further incubated for 20 min at 37°C, 5% CO₂ to complete the hydrolysis of the dye ester and to allow recovery from the dye loading. The plate was then mounted on the microscope stage for study at 37°C under flowing humidified air containing 5% CO₂. Experiments performed to assess steady state pH i before and after acute acidification with and without inhibitors were performed using BCECF on a microscope geared to measure fluorescence spectrum shifts that occur in excitation spectra. The rapid measurements necessary for the ammonium chloride assays depicted in Fig. 4 were performed using C.SNARF, a dye that shifts its emission spectrum as a function of pH changes. The difference in choice of dye relates to the time necessary to acquire an excitation versus an emission spectrum, which is dictated by the speed for acquisition of the data. The acquisition time is a function of the type of light detector used to transduce the fluorescence signal and of the velocity and length of time that is necessary to drive an excitation versus an emission detector through a large number of wavelengths in the amount of time necessary to detect the changes in pH i within the critical time frame in which they occur.

**Inhibitors.** Cariporide was freshly dissolved in deionized distilled H₂O at a concentration of 3 mM immediately before each experiment, then diluted to a final concentration of 60 μM in medium. DIDS was dissolved in DMSO at 0.2 M, and then diluted to 0.2 mM in α-MEM. The CNCn was dissolved in DMSO as a 500-mM stock and then was diluted in α-MEM medium to a final concentration of 10 mM in buffer or media, and then pH adjusted with NaOH to the experimental pH. This was then diluted to a final concentration of 10 mM on the microscope stage. The pH was stable for the duration of all of the experiments, which was confirmed with direct measurements of pH e. DMSO and other vehicle controls were completely negative with respect to the effect on pH e (data not shown). DBDS was made up as a 0.02 M stock solution and diluted to a final concentration of 0.3 mM. The results of dose-response experiments indicated which drug concentrations would give maximum pH i effects (data not shown). For inhibitors that did not show activity, the concentrations used were similar to those that gave maximum responses in other cell lines tested in previous work (5).

**Fluorescence Microscopy and Calibration.** Cellular pH values were obtained from whole emission or excitation spectra that were measured on cells that remained attached to physiological substrate as described above. Samples on an inverted microscope (Nikon Diaphot; Nikon Inc.) were illuminated via fiber optic with a 75-W xenon light source equipped with a spectrofluorimeter that made 140 single-wavelength measurements at 1-nm intervals, from 400 to 540 nm, to cover the entire excitation spectrum for BCECF. The excitation monochromator slit was 4 nm, and each reading had a 0.4 s/step integration time. The emission was detected using a filter cube containing a 510-nm dichroic mirror and a 520-nm emission interference filter (10-nm band width; Omega Optical, Brattleboro, VT) at 550 nm.

Another inverted microscope (Nikon Diaphot; Nikon Inc.) was used to collect emission spectra (for C.SNARF), as has been previously described (5, 28, 30, 31). Briefly, a fiber optic from a 75-W xenon lamp excited the dye at 510 nm., selected by monochromator. A filter cube transmitted the emission wavelengths above 540 nm to a spectrometer that projected the emission spectrum onto a CCD array. The resolution was slightly better than 1 nm/pixel of the CCD. All of the wavelengths of the emission spectrum were measured simultaneously, with light being collected for 5 s and the process repeated every 10 s. This system was used when rapid measurements were desired, as in the ammonium chloride assay (see Fig. 4).

In both cases, the use of the entire spectrum to determine pH is an improvement on the usual ratiometric dye technique, in part because 140 different measurements are made on each cell. This redundancy in the data results in a significant improvement in accuracy. The value of pH i is obtained from a least-squares calculation that involves a form of averaging. Possibly, it is more important that background fluorescence (caused by phenol red and serum in the medium and light-scattering artifacts from cells) was measured on dye-free cells and then subtracted before the calculation was performed (31). In part because the microscope did not have a long working-distance objective, the fluorescence from the medium surrounding the cells was sufficiently small (and constant) that it could be subtracted from the data spectra. There was no detectable change in results when medium containing serum and phenol red was used, as compared with medium that was prepared without serum or the indicator (data not shown). In addition, there was no effect of matrix proteins on basis spectra or baselines.

All of the determinations were made on attached cells at 37°C in medium containing HCO₃⁻ under flowing 5% CO₂ to avoid underestimating Cl⁻/HCO₃⁻ exchange (13). Assays determining NHE-1 activity and MCT activity were made under conditions optimal for maximum activity, i.e., in the presence of serum (32), integrin engagement (33, 34), and during acute and/or chronic acidification (9, 16).

To make a single pH i measurement, the spectrometer was stepped through all excitation wavelengths (in the case of BCECF), or else, the CCD on the emission spectrometer simultaneously acquired a signal for every wavelength in the emission spectrum (as for C.SNARF). A background spectrum was obtained from an identical plate of the same cells under identical conditions, except the cells were not loaded with fluorescent dye. The background spectrum was then subtracted from the data spectrum to obtain the signal that was attributable only to the dye itself. This spectrum was analyzed by a least-squares calculation to find the value of pH i that was most consistent with the data.
Although the background spectrum was measured each day (in case cells should vary in light-scattering properties from one day to the next), the calibration of the instrument was performed only when a change was made in optical elements (adjustment of lamp, change of slit width, and so forth). A new calibration was also done whenever cells of a new type were being tested, but it was found that a change in cell type mainly affected the background spectrum but had little effect on instrument calibration. For example, it was expected that light scattering and other factors might be different enough to cause a problem when tissue fragments were analyzed. Accordingly, the system was recalibrated (see “Xenograft Microfragments” below) for xenograft fragments, but the calibration was found to be the same for cultured or tissue fragment DB-1 cells.

The calibration procedure for adherent cells containing either C.SNARF-1 or BCECF was essentially the same as that described by Grynkiewicz et al. (35) for fura-2, which is another “dual wavelength” ratiometric dye that is used to measure intracellular free calcium; however, fluorescence signals were measured at ~140 wavelengths rather than at only 2. In following the methods of Grynkiewicz et al. (35), cells were placed in a cytosol-like buffer and exposed to the ionophores nigericin and valinomycin to obtain a known pH, i.e., equal to pH e; Ref. 32). As in Grynkiewicz et al. (35), measuring the exact same visual field of cells at two different known values of pH (achieved by modifying buffer pH while on the microscope stage) allowed the computer program to calculate the fluorescence for the dye if it were in the fully protonated or fully deprotonated state. This produced two “basis spectra,” corresponding to the signal from intracellular dye at high pH (fully deprotonated) or low pH (fully protonated; Refs. 5, 30, 36). The computer program that analyzed each sample asked what mixture of these two basis spectra would produce the least-squares best fit to the observed spectrum (31, 36, 37). The relative amounts of protonated- or deprotonated-basis spectra needed revealed the relative amounts of protonated and deprotonated dye in the cells. This information, and the Henderson Haseibelbalch equation, together with the pHs of the dye, provided the pH inside the cells.

The calibration of the system (a particular set of basis spectra) was tested by analyzing cells in which the pH was known because the cells were in cytosol-like buffer and exposed to nigericin and valinomycin (35). Measurements were made on several cell preparations over a period of several days and covering a range of pH values between 6.55 and 7.47. When the calculated pH values were compared with the expected values based on pH e, the accuracy of the system was found to be within 0.01 pH units on some days and within 0.024 units, overall.

Protocol for Extended pH i Time Courses. Dye-loaded cells in medium were mounted on the microscope stage, and steady-state values for pH were determined in 2 ml of media. The cells were then incubated with one or more inhibitors of pH regulation at pH e 6.5, 6.7, or 7.3 for 1–2 h to measure the effects on pH i in a solution. Results from such experiments were always compared with experiments in which pH e was reduced in the absence of inhibitors of pH regulation. The inhibitors included: 0.2 mM DIDS, which preferentially inhibited the Na+-dependent Cl-/HCO3- exchanger (13, 14); 60 μM cariporide, which preferentially inhibited NHE-1 (12); and 10 mM NCN, which preferentially inhibited H+-linked mono- carbamate transporters, isoforms 1, 2, and 4 (see Table 1). These isoforms have various relative affinities for pyruvate and lactate as substrates (16, 18, 38, 39).

The experiments were performed in complete growth medium at 37°C, and cells remained adherent to the substrate to which they had attached during the previous 48–72 h. In each procedure, the first measurement was of steady-state pH i in cells that were dye-loaded in growth medium at pH e 6.7 or 7.3. The initial steady-state pH i at pH e 6.7 or 7.3 was measured several times on a field of 8–15 cells, and then the medium was replaced with medium containing inhibitors and/or a low pH e. The pH i was measured on a single field of cells three to five times every 15 min for 1 h. For long time-course experiments, new plates of cells that were freshly loaded with dye were provided every hour, and the experiment was repeated (as shown in Fig. 1 a) using with one plate for 2 h, and no differences were observed between the approaches. Control experiments showed that the inhibitor solvents (vehicles) had no effect on pH i.

Recovery from an Intracellular Proton Loading. NH4Cl intracellular proton loading protocols were performed on substrate-attached cells in Hanks’ buffer [HEPES, 4 mM NaHCO3 (pH 7.3)] at 37°C as described previously (28, 40). The NH4Cl procedure exposed cells to an intracellular acidification at a normal pH e. The time course of the cell recovery allowed observation of proton transporters when they have been stimulated to a maximum degree. Rates of proton extrusion could then be compared between cell lines. During the recovery from cytosolic acidification, the rate of change, dPH i/dt, was obtained from the slope of the graph of pH i as a function of time. Groups of three data points were used to calculate each rate.

Recovery was monitored in the presence and absence of inhibitors to determine which transporters were most active under conditions of a large intracellular proton load. This differed from exposing cells to inhibitors in the presence of extracellular acidification, which only indirectly caused intracellular acidification.

To test the melanoma cell lines during an intracellular acidification, a particular field of 10–20 cells loaded with C.SNARF were placed on the microscope stage. They were then incubated with 50 mM NH4Cl for 350 s. Three to five spectra were taken at 10-s intervals. Next, the buffer was changed to an NH4Cl-free buffer, still at pH e 7.3. The removal of NH4Cl induced a rapid acidification as NH4 exited the cell (40, 41). C.SNARF fluorescence was excited at 510 nm, and the recovery of cells from intracellular proton loading was monitored for 350 s or longer if recovery was incomplete. Every time point for a given experiment contained the average (± SD) of three to five measurements taken over a field. Each experiment was repeated at least twice, and the data from a representative experiment is presented as pH i (± SD) versus time.

Experimental Approach/Statistics. Three to five consecutive spectra were collected every 15 min, and were
averaged to calculate a pH\textsubscript{i} value for each 15-min interval with a SD. The error bars are not visible on the graphs when they are smaller than the symbols. Each experiment was repeated three or more times, and data from representative experiments are depicted in the Figures. All of the experiments were performed at least three times, but only one is shown in each case instead of averages because the initial pH\textsubscript{i} varied slightly from plate to plate for all of the melanoma cell lines, although the magnitude of the pH\textsubscript{i} changes in the case of various treatments did not. This was attributed to the clonal variation of the melanoma cells, because it was not observed in the nonmalignant and other malignant lines that were analyzed in the present study and in previously published work from this laboratory. This variation did not occur within one plate of cells when numerous fields were analyzed; rather, it was interplate variation.

**Flow Cytometry.** These experiments were performed using a CALTAG kit (Burlingame, CA) according to the manufacturer’s instructions. Antibodies were polyclonal and raised in rabbits against the human antigens. NHE-1 antibody was obtained from Chemicon (Temecula, CA), and MCT-1 and MCT-2 antibody were obtained from David Golde (Memorial Sloan Kettering, New York, NY). Antibodies to MCT-1 and MCT-4 were also obtained from Andrew Halestrap (University of Bristol, Bristol, England). Briefly, cells were grown on fibronectin/collagen I substrate for 2 days, then released using Cell Stripper (Mediatech, Herndon, VA), a nonenzymatic agent, and rinsed in three changes of sterile, autoclaved, PBS buffer adjusted to the growth pH of the cells (6.7 or 7.3). Cells were counted and then resuspended at a concentration of 5 × 10^{5} cells/250 μl in FACS medium [PBS, 0.1% sodium azide, 1% BSA (pH 7.0)]. All of the antibodies used in these experiments were directed toward the COOH-terminal ends of the transporters, located in the intracellular portion of the proteins, and, therefore, detergent permeabilization was necessary in each experiment to ensure primary antibody contact with the protein. Plates were centrifuged for 3 min at 1500 rpm, and the supernatant was removed. The 50 μl of reagent A (fixation medium) was added to each well and incubated for 15 min, followed by three washes in FACS medium. Cells were then incubated with primary antibody at room temperature for 45 min, washed, and then incubated with secondary antibody at room temperature for 45 min. Controls included cells incubated either with secondary antibody alone, or with no antibody, to measure nonspecific fluorescence and background fluorescence, respectively. Other nonmelanoma cell lines were also analyzed for comparison. Cells were monitored using a FACSCalibur flow cytometer (Becton Dickinson, Franklin Lakes, NJ) equipped with CellQuest software. Data are expressed as relative fluorescence.

**RT-PCR.** RNA was isolated from cells grown at normal and low pH after 2-days growth on a collagen I/fibronectin substrate. The purity and quantity of the RNA was assessed by measuring its absorbance at 260 nm and 280 nm and determining that the ratio of these absorbances was 1.5–2.0; a total protein-obtained harvest (20 million per harvest from each pair of T25 flasks) was 50 μg. Cells were detached from flasks using Cell Stripper (Mediatech, Herndon, VA) and then

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**Fig. 1. a,** the effect of single inhibitors on Sk-Mel-28 cells at their steady-state pH\textsubscript{i} of 7.3. pH\textsubscript{i} is shown as a function time for Sk-Mel-28 melanoma cells with no drug additions (○), or with 0.06 mM cariporide (●), with 0.2 mM DIDS (▲), or 10 mM CN\textsubscript{5} (▼) added immediately after 0 time measurements were obtained. **b,** DB-1 cells at their growth pH\textsubscript{i} without (○) and with (●) cariporide; acute extracellular acidification to pH\textsubscript{e} 6.7 (□) and with cariporide (●); and acute extracellular acidification to pH\textsubscript{e} 6.5 (▲) and with cariporide (▲). Each graph is representative of two or more similar experiments performed on separate days. Values and error bars, means ± SDs. c, DB-1 cells at their growth pH\textsubscript{e} without (○) and with (●) cariporide, acute extracellular acidification to pH\textsubscript{e} 6.7 (□) and with cariporide (□); and acute extracellular acidification to pH\textsubscript{e} 6.5 (▲) and with cariporide (▲). Each graph is representative of two or more similar experiments performed on separate days. Values and error bars, means ± SDs. Steady-state measurements were performed using cells containing BCECF using an excitation monochromator and a photon-counting detector (see also Figs. 2 and 3).
were centrifuged in sterile PBS adjusted to the growth pH. RNA was prepared using RNAzol, as per instructions from Tel-Test Inc. (Friendswood, TX). The purity and quality of the RNA was determined by the appearance on a 1% agarose gel. The amount of RNA was quantitated using an Alphalmager 2200 equipped with the v5.5 program (Alpha Innotech Corp., San Leandro, CA), by comparison with a 100-bp nucleotide ladder (Promega, Madison, WI). cDNA was synthesized from total RNA using Superscript II RNase H Reverse Transcriptase (Life Technologies, Inc., Rockville, MD) using random hexamer primers (Boehringer Mannheim, Indianapolis, IN) in the presence of 0.25 μl RNasin, (40 units/μl; Promega, Madison, WI). Reverse transcription was carried out in an MJ Research, Inc. thermal cycler using a 60-min incubation at 42°C followed by 10 min at 50°C and 15 min at 70°C. The reaction was then cooled to 4°C and the random hexamers were removed using a QIAquick PCR purification kit (Qiagen Inc., Valencia, CA).

For the PCR reaction, 2.0 μl cDNA, 2.0 μl Aquanase (Tel Test Inc., Friendswood, TX), and 2.5 μl each of 10 μM forward and reverse primers were combined and heated for 2 min at 95°C in an MJ Research, Inc. thermal cycler and then cooled rapidly on ice. A master mix of 31.5 μl Aquanase, 5.0 μl of 10× PCR buffer, 2.0 μl 10 μM dNTPs (Amersham Pharmacia Biotech, Piscataway, NJ), and 2.5 μl Sigma Red Taq (Sigma) was added to the cDNA mixture. Cycle times were as follows: 2 min at 92°C; 35 cycles of 30 s at 92°C, 1 min at 50°C or 60°C; followed by 2 min at 72°C, 72°C, 5 min, 4°C, end), using transporter isof orm-specific primers. The primers were obtained from Life Technologies, Inc. and were designed based on published sequences. An annealing temperature of 50°C was used for the MCT-1, MCT-4, and HUGAPDH208 primers, and 60°C for the NHE-1 and HUGAPDH325 primers. PCR samples were prepared using identical starting quantities of mRNA, using GAPDH message levels as an internal standard. PCR products were analyzed by size fractionation in a 1.4% agarose gel with a concentration of 0.5 μg/ml ethidium bromide (Life Technologies, Inc.).

**Xenograft Microfragments.** Fresh explants of human DB-1 tumor tissue (1–2 mm each) were excised from a SCID mouse on each of three occasions. SCID mice were anesthetized and tumor pH was measured with a 20-gauge needle combination microelectrode in vivo before sacrifice. The average pH of DB-1 mouse tumor xenografts were 6.68 ± 0.05. Samples of the tumor were excised and kept on ice in medium at pH 6.7 with 10% serum and 10 mM 2-[N-morpholino]-ethanesulfonic acid (to maintain pH 6.7 outside of an incubator with a 5% CO₂ atmosphere), dissected with a scalpel, and sieved through a metal mesh grid to divide them into small fragments between 10–20 cells in diameter. The microfragments were then affixed to microwell dishes using Cell-Tak tissue adhesive (Collaborative Biomedical Products, Bedford, MA) which had previously been layered onto microwell plates. Cells affixed to Cell-Tak were incubated for about 4 h at a pH that was the same as the measured pH in situ immediately before tumor excision (6.7). Other microfragments were incubated at 6.5 and 6.3. These were compared with microfragments incubated at the same pH in the presence of 10 μM CNCn.

**Results**

In steady state, pH of cells was first measured at time zero in complete medium at the growth pH of the cells. In control runs, the cells and medium were monitored with no further perturbation. The effects of inhibitors on pH were then tested by incubating cells with inhibitors, either at pH 7.3 or during extracellular acidification to 6.7 or 6.5 after the initial steady-state pH was taken at time zero. Whenever the pH decrease in the presence of inhibitor was significantly greater than that caused by acute acidification alone, it indicated that a particular transporter was active in that cell type.

**The Effect of Ion Transport Inhibitors at pH 7.3.** After a steady-state measurement was made, cells were exposed to 60 μM cariporide, 0.2 mM DIDS, or 10 mM CNCn, and pH was monitored for an additional 1–2 h. All three of the cell lines were tested with all three of the inhibitors at pH 7.3. Under these conditions, an effect on pH was observed in only one instance. Sk-Mel-28 cells in the presence of DIDS exhibited a decrease in pH of about 0.2 unit within 1 h. No detectable change in pH occurred in the presence of either cariporide or CNCn for these cells (Fig. 1a). Similarly, no detectable effects were seen after administration of any of the inhibitors in DB-1 and DB-8 melanoma cells at pH 7.3 (data not shown). These results were not unexpected, because the Cl⁻/HCO₃⁻ exchanger had previously been reported in other cells to be the only transporter active at pH 7.3 (13).

**The Effect of Ion Transport Inhibitors When Combined with Acute Acidification to 6.7 or 6.5.** Following the experience in other cell types in which NHE-1 activity was the cell’s prime defense against extracellular acidification (13), this was what was tested first in the melanoma cells. In Fig. 1b, dashed lines (controls, with open symbols) show the effect of various pH on pH in DB-1 cells in the absence of any inhibitor. The solid lines with filled symbols show the additional effect of cariporide at each pH tested. It was striking that cariporide did not lower pH under conditions that were previously found to be optimal for inhibiting NHE-1 activity in other cell types (12), which included acidification to pH 6.7 or 6.5. Similar results were obtained in experiments on Sk-Mel-28 and DB-8 melanoma cells (data not shown).

Fig. 1c depicts an evaluation of a Cl⁻/HCO₃⁻ exchange inhibitor during acidification on pH. The dashed lines with open symbols show the effect of acute acidification to pH 6.7 and 6.5 alone on DB-1 cells (without inhibitor). The solid lines with filled symbols show that there was no effect of DIDS at pH 7.3, 6.7, or 6.5. Similar results were obtained using DB-8 cells (data not shown).

Fig. 2a depicts an evaluation of MCT activity using 10 μM CNCn in DB-1 melanoma cells with different degrees of extracellular acidification. As previously stated, there was a negligible effect of CNCn at pH 7.3 (open versus filled circles). Acidification alone, from pH 7.3 to 6.7, reduced pH by 0.30 units (open squares) from 7.23 ± 0.03 to 6.93 ± 0.01. With the addition of CNCn, the pH further decreased to 6.70 ± 0.02; a total acidification of −0.53 units (filled squares). Acidification to pH 6.5 lowered pH to 6.66 ± 0.01, a change of −0.57 units (open triangles), and the inclusion of CNCn lowered it further, to 6.23 ± 0.03 a total of −1.00 pH...
The effect of acute acidification to pH 6.5 or 6.3 with CNCn on DB-1 melanoma experiments performed on separate occasions. Values formed on separate days. Values Each graph is representative of two or more similar experiments performed on separate occasions. Fig. 2. a, the effect of acute acidification to pH 6.7 or 6.5 with CNCn on DB-1 melanoma cells. DB-1 cells at their growth pH without ($\bullet$) and with (●) CNCn, acute extracellular acidification to pH 6.7 ($\square$) and with CNCn ($\diamond$), and acute extracellular acidification to pH 6.5 ($\triangle$) and with CNCn ($\square$). Each graph is representative of two or more similar experiments performed on separate days. Values and error bars, means ± SDs. b, the effect of acute acidification to pH 6.5 or 6.3 with CNCn on DB-1 melanoma cells at their growth pH 6.7 without (●) and with (●) CNCn and with cariporide (gray filled circles), acute extracellular acidification to pH 6.5 ($\square$) and with CNCn ($\square$), and acute extracellular acidification to pH 6.3 ($\triangle$) and with CNCn ($\square$). Each graph is representative of two or more similar experiments performed on separate occasions. Values and error bars, means ± SDs.

pH 6.5 caused pH$_i$ to decrease to 6.54 ± 0.01 and 6.67 ± 0.01, respectively (data not shown). These effects represented changes in pH$_i$ attributable to the drug exposure of 0.06 and 0.14, respectively. The finding with cariporide was consistent with a study elsewhere in which NHEM also exhibited NHE-1 activity (11).

Inhibitors of MCT-1 Acting on DB-1 Cells Adapted to Growth at Low pH. The steady-state pH$_i$ value for DB-1 melanoma cells grown at pH 6.7 and measured at pH 6.7 was ~6.90. Similar values were seen with the other two melanoma cell lines. These were detectably lower than the steady-state values for cells that were grown at pH 7.3 and that were measured also at their growth pH$_i$.

Fig. 2b shows that CNCn had a noticeable effect on the cells grown at pH 6.7, reducing pH$_i$ by about 0.25 unit when they were tested at pH 6.7 (filled black circles). This is comparable with the 0.30-pH-unit effect that CNCn had on cells grown at pH 7.3 when they, too, were tested at pH 6.7 (Fig. 2a).

When the low-pH-adapted cells were acidified to pH 6.5 (Fig. 2b, open squares), the effect that could be attributed to CNCn (filled circles) was more than an additional 0.5 unit. In fact, the pH$_i$ measured <6.0 within 1 h of lowering the pH$_e$. Actual data values below a pH of 6.0 are not reported because limitations of the BCECF dye itself preclude making measurements with precision below that value. Nonetheless, it is clear that the low-pH-adapted cells appeared to be more dependent on MCT-mediated proton removal than did the cells grown at pH 7.3. When the latter were tested at pH 6.5, the effect of CNCn was to reduce pH$_i$ by 0.43 unit (Fig. 2a).

When low-pH-adapted cells were acidified to pH 6.3 (Fig. 2b, open triangles), CNCn (closed triangles) caused a pH reduction to below pH 6.0 within 30 min. Although pH values cannot be stated with precision in this range, it was clear that, for an additional 60 min, pH$_i$ remained below 6.0, and the cells remained intact in the sense that the dye was well retained.

An additional MCT inhibitor confirmed the changes in pH$_i$ in all three of the cell lines tested. This was DBDS, which is primarily an MCT-1 inhibitor and which has the advantage of exclusion from the cell interior because of highly charged sulfonic moieties that cause a lack of ability to enter into cells and negligible effects on mitochondrial monocarboxylic acid transport as a result (43). In Fig. 3, the effects of DBDS on DB-1 cells grown at pH 7.3 and monitored at pH 7.3, 6.7, or 6.5 are shown. The results are similar to those obtained using CNCn. DBDS did not affect pH$_i$ at pH 7.3 (open and filled circles) but did cause pH$_i$ to decrease when combined with acidification to pH 6.7 (open and filled squares), and even more so with acidification to pH 6.5 (open and filled triangles).

Recovery from Intracellular Acidification in Melanoma Cells, Breast Cells, and Keratinocytes and Assessment of Buffering Capacity. Fig. 4a shows the patterns of recovery of pH$_i$ that have typically been observed in other cell
types (HMT3522-T4/2 malignant breast cells in this case). The recovery rate (expressed as $d\text{pHi}/dt$, the slope of the graph shown) reached $0.25 \pm 0.64$ ($n = 4$) pH unit/min ($\text{open circles}$). These cells fully returned to their initial steady-state pH$_i$ within 350 s after the removal of ammonium chloride (ammonium chloride challenge itself is not shown in Fig. 4).

When NHE-1 was inhibited by cariporide, the recovery rate was reduced to $0.06 \pm 0.02$ ($n = 4$) pH unit/min (Fig. 4, $\text{inverted filled triangles}$), and cells did not return to their initial pH$_i$. Recovery was not affected by DIDS ($\text{filled triangles}$) or CNCn ($\text{filled squares}$), which indicated that the Cl$^-$/HCO$_3^-$ exchanger and MCTs did not figure prominently in this recovery. The nonmalignant counterpart of HMT3522-T4/2, called HMT3522-S1 (44), exhibited identical recovery profile and responses to inhibitors (data not shown). In a similar experiment, recovery from intracellular acidification was measured in keratinocytes (HaCaT cells) with inhibitors present or absent during the recovery (Fig. 4b). The rate of recovery was $0.11 \pm 0.03$ pH unit/min ($\text{open circles}$), which was $\sim100\%$ inhibited by 60 $\mu$M cariporide ($\text{filled inverted triangles}$), to a rate of $0.007 \pm 0.002$ pH unit/min ($n = 4$). Again, recovery was not affected by DIDS ($\text{filled triangles}$) or CNCn ($\text{filled squares}$).

In contrast to all three of these cell lines, Sk-Mel-28 cells, grown at pH$_i$ 7.3, depicted in Fig. 4c, had a recovery during resuspension that was slow and incomplete, at an average rate of $0.06 \pm 0.01$ pH unit/min ($n = 4$). DB-1 and DB-8 cells had similar profiles (data not shown). These rates are representative of at least four experiments each. The filled symbols show inhibitor effects, which were only barely discernable with CNCn and not at all discernable with DIDS and cariporide.

For cells grown at pH$_i$ 7.3, the final values of pH$_i$ that were reached for both of the melanoma cell lines were less than 6.8, although both cell lines had original steady-state pH$_i$ values between 7.20 and 7.30. Experiments were also performed monitoring recovery for an additional 350 s or incubating cells after treatment and measuring pH$_i$ after 24 h. No further recovery of pH$_i$ was observed (data not shown). These data are similar to results obtained by two other in-
investigators using a total of eight other melanoma cell lines (11, 45).

Flow Cytometry Using Antibodies to MCT-1, MCT-4, NHE-1. Fig. 5, a and b, show that MCT-1 and MCT-4 proteins were present in DB-1 cells grown at pH 7.3 (filled peaks), and may be enriched in melanoma cells grown at pH 6.7 (Fig. 5, c and d, open peaks). In this experiment, the levels of MCT-4 were elevated 2- to 3-fold, and MCT-1 was elevated 10-fold in DB-1 cells grown at pH 6.7 as compared with those grown at pH 7.3. In Sk-Mel-28 cells, the elevations of bound antibody were less reproducible. DB-8 cells had high levels of both MCT-1 and MCT-4 even in cells grown at 7.3, and showed correspondingly less enhancement under low pH conditions. All three of the melanoma cell lines also had detectable NHE-1 protein, shown elsewhere in five other melanoma cell lines (11). The present study showed that NHE-1 levels were unaffected by growth at low pH in all of the three cell lines (data not shown). The results are consistent with the possibility that up-regulation of MCTs may have been an important mechanism by which these three melanoma cell lines maintained pH, while being grown at a low pH. MCT-1 levels in melanoma cells grown at pH 7.3 were similar to those observed in HaCaT and HUVEC (Fig. 5, c and d, open peak, solid line), but MCT-4 levels were higher in melanoma cells (open peak, dashed line). The flow cytometric data were consistent with the inhibitor data, which indicated that MCTs are up-regulated, with changes in expression levels being clearly reproducible in DB-1 cells and slightly less consistent in Sk-Mel-28 cells. In DB-8 cells the levels of MCT were high when cells were grown at either pHo.
6.3, CNCn caused pHi to decrease to values below pHi 6.0. This decrease was sustained for 1 h without dye leakage. The results show that CNCn could significantly compromise pHi in a xenograft, in which cells in situ retain the cell-cell contacts.

Discussion

Three families of transporters that are found in the plasma membrane are designed to maintain pHi in the viable range. Although many cells are known to rely on all three families, the melanoma cells in this study were uniquely dependent on the MCT family. MCTs, most commonly known in the literature as “lactate-proton symporters,” remove both lactate and H+ (in the same direction). The MCT could be passive in that no other reaction is coupled to serve as a source of energy (e.g., no hydrolysis of ATP or transport of Na+). Other ion gradients are involved besides the gradients of lactate anions and the H+ themselves. In a tumor with an acidic milieu, energy stored in a lactate gradient could be used to drive H+ transport against its transmembrane gradient, as originally described by Warburg (46). That the MCT would be the predominant proton-removal mechanism for any cells, and especially for cells in an extracellular medium of reduced pHi, was unexpected, given that the combined lactate and H+ gradients provide the only free energy that is available for the joint transport.

Changes in pHi, as well as the regulation of pHi, have been studied in a number of tumor types. Activities of transporters that affect pHi have been connected to the cell cycle (32, 47–49), cell migration (50), and drug resistance (51), and to potential for metastasis (45, 52). In many solid tumors in which pHi is lower than normal, individual cell survival requires that cells have development mechanisms for the control of pHi that can remain active despite the increase in the free energy that is required to remove metabolic protons from the interior of the cell. Because tumor cells are often found at a low pHeo, even one that would normally compromise the viability of cells, it is of interest to understand the mechanisms by which these cells cope with the challenge of increased extracellular proton concentration. Selective inhibition of such mechanisms can also be explored as a way to increase stress on tumor cells and aid, for example, in chemotherapy or hyperthermia treatment.

Melanoma is a notoriously heterogeneous disease in terms of phenotypic and genotypic characteristics (53) and of response to therapy (54). It is, therefore, noteworthy that the three melanoma cell lines examined were strikingly similar with respect to pHi regulation. The data are suggestive that Sk-Mel-28 cells had two of the three major pHi-regulating transporters. At a growth pHeo of 7.3, they primarily used the Cl−/HCO3− exchanger. The MCTs were induced by extracellular acidification, but there was little NHE-1 activity even at low pHeo. DB-1 and DB-8 cells also exhibited little NHE-1 activity and significant MCT activity, and these also lacked measurable Cl−/HCO3− exchange.

In an acidified medium, H+ removal via the MCT would require an increase in intracellular lactate and H+ levels to force flow in the outward direction. It is not known why melanoma cells rely on this mechanism rather than on NHE-1, which could draw on the energy of the sodium gradient and would be expected a priori to be more effective at transporting protons against a greater proton gradient in acidified environments. Such a dependence on the MCT has not been observed in any of the normal cells or other tumor cell types tested in this laboratory, nor has it been reported elsewhere. Interestingly, it also contrasts with keratinocytes (which typically surround melanomas) and melanocytes (the cells of origin). Both of these showed typical NHE-1 activity.

In the intracellular acidification challenge assay, melanoma cells did not recover rapidly or fully, even in control experiments without inhibitors present. Cells are usually thought to use the NHE-1 to recover from intracellular acidification, so their poor recovery during this assay is additional evidence that NHE-1 activity is low in these melanoma cell lines. Similar results from assays testing NHE-1 function have been reported in a total of nine other melanoma cell lines by two different investigators (11, 45). Our data in the present study, for three other melanoma cell lines and two other melanocyte lines, are consistent with those data.

Many previous studies using CNCn or quercetin have focused on lactate transport (16, 38, 39, 55, 56), but none have directly measured concomitant effects on pHi. MCT-1 gene expression has been studied in some tumors (57) exclusive of melanoma, and MCT-4 levels have not been measured previously in any tumors. This is, therefore, the first demonstration that the MCTs appear to be critical for maintaining pHi in the viable range in any cell type. It was also necessary to study tumor cells under low pHeo conditions for the MCT activity to be manifest. This work indicates that inhibitors of MCTs could be useful adjuncts to melanoma therapy, with particularly high activity in areas of tumors in which pHeo is low, giving them a built-in selectivity for the tumor bed. It is because of their low NHE-1 activity that inhibitors of MCTs would be expected to be particularly effective in melanoma as opposed to other tumor types.

One caveat with respect to the experiments using CNCn (18) is that CNCn may also act on mitochondria to inhibit pyruvate entry via a MCT variant, which could cause increased lactate and H+ production. This secondary effect could contribute to the decrease in pHi observed in experiments with CNCn because the inhibited MCTs might be trying to keep up with the increased proton production. Two lines of evidence suggest that metabolic shifts involving increased lactate and proton production are not the principal reason for the observed decreases in pHi. First, measurements of lactate production when the extracellular milieu was acidified found it to be reduced, rather than increased (42). Second, when the inhibitor DBDS was used instead of
CNCn, it produced the same results. The sulfonate side chains of DBDS have been reported to preclude its entry into cells, with the result that it acts only on the plasma membrane MCT (43). Other considerations that pertain in vivo are the level of hypoxia and blood flow levels, which are not addressed in the present study.

In conclusion, the data presented here indicate that MCTs are the major functional transporters in melanoma cells. Inhibitors of MCTs have a great potential to selectively compromise melanoma cell viability and to improve the effectiveness of chemotherapeutic drugs that are more effective at low pHp/Ha, such as alkylating agents and platinum-containing compounds, while sparing normal and surrounding cells that are not in regions of low pHa.

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References

Regulation of Intracellular pH in Human Melanoma: Potential Therapeutic Implications

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